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CBSE Class 12 Chemistry Notes: Solutions - Basics ... Solutions Class 12 Notes Chapter 2. A solution comprises a solute and a solvent. It is defined as a homogeneous mixture of two or more than two substances. They can be classified into three types: Solid solutions. Gaseous solutions. Liquid solutions. Molarity, mole fraction, percentages, and molality are the terms used to express the concentration of a solution.

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Class 12 Chemistry Revision Notes for Chapter 2 - Solutions Revision Notes on Solution: A solution is a homogeneous mixture of two (or more) substances, the composition of which may vary between certain limits. A solution consisting of two components is called binary solution. The component which is present in large quantity is called solvent and the component which is small in quantity is called solute. If both components are in same physical state.

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