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A chemical bond is formed when electrons are shared between two atoms. There are three types of bonds: covalent bonds, polar covalent

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bonds and ionic bonds.

The simplest example of bonding is demonstrated by the  $H_2$  molecule. We can see that each hydrogen atom has a single electron from the periodic table.

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7.2 IONIC BONDING

The chemical bond

formed by transfer of

electron from a metal to a

non- metal is known as

ionic or electrovalent

bond . For example,

when sodium metal and

chlorine gas are brought

into contact, they react

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violently and we obtain sodium chloride. This reaction is shown below:



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## 7 CHEMICAL BONDING

There are many types of chemical bonds that can form, however the 3 main types are: ionic, covalent, and metallic bonds. You must

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become familiar with how they work and the differences between the 3 types. I. Ionic bonding: Model 1 is a description of what chemists call ionic bonding. Ionic bonding occurs strictly between metal and nonmetal atoms.

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Accordingly, there are  
different types of  
chemical bonds, like,  
Ionic or electrovalent  
bond Covalent bond Co-  
ordinate covalent bond  
In addition to these we  
have a special kind of

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bond called hydrogen bond. Let us discuss about different types of bonds, their formation and the properties of the compounds so formed.

5.2 Ionic or Electrovalent Bond

## CHEMICAL BONDING

There are two main types and some secondary types of chemical bonds:

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1 Ionic bond Ionic bonding involves a transfer of an electron, so one atom gains an electron while one atom loses an electron. One of the resulting ions carries a negative charge (anion), and the other ion carries a positive charge (cation).

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which a sodium atom

gives an electron to a

chlorine atom Force that

causes a ...

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