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Author: dedfsf dgdgfdgd, Name: Student  
Exploration- Balancing Chemical Equations  
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## Student Exploration Balancing Chemical Equations (ANSWER... Answer Key

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ . 1. How many atoms are in a hydrogen molecule ( $\text{H}_2$ )? 2. How many atoms are in an oxygen molecule ( $\text{O}_2$ )? 3. How many hydrogen and oxygen atoms are in a water molecule ( $\text{H}_2\text{O}$ )?

## Balancing Chemical Equations Student Exploration Gizmo Lab...

The two chemical equations each have an arrow in them. The arrow represents a chemical change. Use the pieces and equations to try and balance each of the formulas so that there is the same number of atoms on each side of the chemical reaction. Use the labels to label the reactants and

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Chemical Equations  
products on your 2 equations.

## Answer Key

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Student Exploration: Balancing Chemical  
Equations. DEFINE Vocabulary:

coefficient. Combination- a chemical  
reaction when two reactions combine to  
make one. Compound- a substance  
consisting of two or more elements .

Decomposition- the separation of a  
substance into a simpler form . Double  
replacement- multiple elements that are  
getting swapped

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The equation  $H_2 + O_2 \rightarrow H_2O$  is unbalanced  
because there are two oxygen atoms on the  
reactants side of the equation, and only one  
on the products. side of the equation. To  
balance the equation, you cannot change the  
structure of any of the molecules, but you

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can change the number of molecules that are used.

## Answer Key

### ~~Student Exploration Sheet: Growing Plants~~

This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction:  $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}$ . How many atoms are in a hydrogen molecule ( $\text{H}_2$ )?

### ~~Balancing Equations Gizmo—Student Exploration Balancing ...~~

Chemical Equations. Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Balancing Chemical Equations. Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data. **DOWNLOAD.**

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**DESCRIPTION.** Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate

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## Answer Key

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The law of conservation of matter states that  
no atoms are created or destroyed in a  
chemical reaction. Therefore, a balanced  
chemical equation will show the same  
number of each type of atom on each side of  
the equation. To set up an equation in the



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Chemical Equations Gizmo™, type the chemical formulas into the text boxes of the Gizmo

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Student Exploration: Balancing Chemical Equations In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo™, look at the floating molecules below the initial reaction:  $H_2 + O_2 \rightarrow H_2O$ . How many atoms are in a hydrogen molecule ( $H_2$ )?

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Balancing Chemical Equations. Balance and classify five types of chemical reactions: synthesis, decomposition, single replacement, double replacement, and

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Chemical Equations  
Answer Key

combustion. While balancing the reactions, the number of atoms on each side is presented as visual, histogram, and numerical data. Use for 5 minutes a day.

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Chemical Equations Gizmo Answer Key  
The first step to balance the equation is to  
write down the chemical formula of  
reactants that are listed on Page 23/30  
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Balancing Chemical Equationsthe left side of  
the chemical equation.

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Equations~~

To balance a chemical equation, you first  
need to be able to count how many atoms of  
each element are on each side of the

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equation. In this activity, you will practice counting the atoms that are represented in chemical formulas.

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