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Chapter 9 Compressible Flow 647 Solution: For CO<sub>2</sub>, from Table A.4, take  $k = 1.30$  and  $R = 189 \text{ J/kg K}$ . Compute the specific heat:  $c_p = kR / (k - 1) = 1.3(189) / (1.3 - 1) = 819 \text{ J/kg K}$ . The results follow in sequence: (1)  $(1.3 - 1) / 1.3 = 0.154$ . (a)  $2 \times 10^5 \text{ Pa} / (373 \text{ K})(80/125)$ . (a)  $k = 1.3$ .  $T_2 = 336 \text{ K}$  Ans

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