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Chapter 10

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

3 Atoms: The Building Blocks of Matter

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CHAPTER 9 REVIEW Stoichiometry MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Given the following equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$ 4 a. What is the value of the coefficient x in this equation? 40.07 g/mol b.

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What is the molar mass of C₃H₄? 2 mol O 2:1 mol H₂O c. What is the mole ratio ...

mc06se cFMsr i-vi

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Vocabulary from Chapter 6 of the Modern Chemistry textbook. (30 total words) Terms in this set (30) chemical bond. the attractive force that holds atoms or ions together. ionic bonding. a force that attracts electrons from one atom to another, which transforms a neutral atom into an ion.

Modern Chemistry Chapter 6 Vocabulary Flashcards | Quizlet

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