

## Modern Chemistry Chapter 3 Test Answer Key

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Name CHAPTER 3 TEST continued Date Class FILL IN THE BLANK Write the correct term (or terms) in the space provided. 9, If a particular ompound is composed of elements A and B, the ratio of the mass of B to the mass of A will alw ys be the same. This is a statement of the law of exactly 12 g of carbon-12 is referred to as a(n) 11.

~~San Ramon Valley High School~~

CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

Holt McDougal Modern Chemistry 3 Chapter Test Chapter Test B, continued 16 Modern chemistry chapter 3 test b answers. The measure of the ability of an atom in a chemical compound to attract electrons from another atom in the compound is called \_\_\_\_\_. 17. The energy required to remove one electron from an atom is called its \_\_\_\_\_. 18.

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CH 3 Pre Test . CH 3 Nuclear Equations . CH 3 Nuclear Equations II . CH 3 Balancing Nuclear Equations. CH 3 Nuclear Chemistry . CH 3 Pre Test Calculations. CH 3 Protons-Neutrons-Electrons-Mass # -Atomic # CH 3 Atomic #, Mass #, P-N-E Chart II. CH 3 Review Atomic Theory . CH 3 Test

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Modern Chemistry Chapter 3 Review Answers Modern Chemistry Chapter 3 Test. Honors Chemistry Chapter 3 Test. STUDY. PLAY. law of conservation of mass. mass is neither created nor destroyed during ordinary chemical reactions or physical changes  $2H_2 + O_2 = 2H_2O$ . law of definite proportions. Holt Mcdougal Modern Chemistry

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CHAPTER 3 REVIEW Atoms: The Building Blocks of Matter SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Explain the difference between the mass number and the atomic number of a nuclide. Mass number is the total number of protons and neutrons in the nucleus of an isotope.

~~3 Atoms: The Building Blocks of Matter~~

Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. \_\_\_\_\_ 7. A solution containing 35 g of  $Li_2SO_4$  dissolved in 100 g of water is heated from  $10^\circ C$  to  $90^\circ C$ . According to information in the figure, this temperature change would result in a. an additional 5 g of  $Li_2SO_4$  in ...

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Standardized Test Prep CHAPTER REVIEW To give students practice under more realistic testing conditions, give them 60 minutes to answer all of the questions in this Standardized Test Preparation Answer the following items on a separate piece of paper. MULTIPLE CHOICE I. In the modern periodic table, elements are arranged according to

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