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SOLUTIONS FOR CHAPTER 21 PROBLEMS 4, 12, 19, 25, 33, 40, 50, 75, 89, 96. 21.4. IDENTIFY: Use the mass m of the ring and the atomic mass M of gold to calculate the number of gold atoms. Each atom has 79 protons and an equal number of electrons. SET UP: 23 NA 6.02 10 atoms/mol. A proton has charge +e.

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