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S-14 1. Solubility of Ethanol in Water Explain why ethanol (CH₃CH₂OH) is more soluble in water than is ethane (CH₃CH₃). Answer Ethanol is polar; ethane is not. The ethanol —OH group can hydrogen-bond with water. 2. Calculation of pH from Hydrogen Ion Concentration What is the pH of a solution that has an H⁺ concentration of (a) 1.75 × 10⁻⁵ mol/L; (b) 6.50 × 10⁻¹⁰ mol/L; (c) 1.0 × 10⁻⁴ mol/L;

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