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6. A spontaneous process is—MCAT 2010 (A) Unidirectional and irreversible (B) Unidirectional and a real process (C) Irreversible

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and a real process (D) All of the above 7. The standard enthalpy of solution of $\text{NH}_4\text{-Cl}$ is $\square\square\text{KJmol}^{-1}$ MCAT 2010 (A) +16.2 (B) -25.0 (C) +4.98 (D) +26.0 8. Reactants have high energy than products in: MCAT 2013

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