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bonds, atoms tend to attain the: A

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state of higher energy; The electron configuration of noble gas atoms; The electron configuration of halogen atoms; All of the above. A metallic bond is a bond between: Valence electrons and positively charged metal ions; The ions of two

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different metals; A metal and nonmetal; None of the above

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In chemistry, we refer to
Chemical bonding as a means or
a way by which an atom attaches
itself with other atoms. Basically,

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there are three types of chemical bonding in chemistry, and they are covalent bonding, ionic bonding, and metallic bonding. In this IGCSE chemistry chemical bonding blog post, I am going to cover the basic concepts of these three types of bonding.

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bonds, atoms tend to attain the: A
state of higher energy; The
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gas atoms; The electron

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configuration of halogen atoms;
All of the above. A metallic bond is a bond between: Valence electrons and positively charged metal ions; The ions of two different metals; A metal and nonmetal; None of the above

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