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ozone formation, c

hlorofluorocarbons,
CFC's 1.2

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Matter Chemistry
Central Science

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Matter and Change

Chapter 1 Section
2 MATTER and Its
Properties Matter -

Anything that has
MASS and VOLUME.

ATOMS and
MOLECULES the
basic building
blocks of matter.

ATOM-- the
smallest unit of an
element that still
maintains the

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properties that
element.

ELEMENT– a pure
substance made up
of only one type of
atom.

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and Change
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Section 3.1

Properties of

Matter pages

70–75 Problem-

Solving Lab 1.

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Explain why the flow of a compressed gas must be controlled for practical and safe use.

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nine 4. accepted 5.
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elements 8. atomic
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Electrons in Atoms

Reviewing

Vocabulary Match
the definition in

Column A with the
term in Column B.

Column A Column

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B 1. The set of frequencies of the electromagnetic waves emitted by the atoms of an element 2. The minimum amount of energy that can be lost or gained by ...

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property. intensive
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properties
(examples)

intensive property
(examples) a
property that
depends on the
amount of matter
in a sample. a
property that
depends on the
type of matter in a
sample not....

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mass, volume,
length, weight.

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number of moles of

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known substance C

H C H C H O O O

CO CO CO H O H O

H O number of

moles of unknown

substance O CO H

O C H CO H O C H

O H O C H O CO

use mole ratio 5

mol O 2 1 mol C 3

H 8 3 mol CO 2 1

mol C 3 H 8 4 mol

H 2 O 1 mol C 3 H

8 1 mol C 3 H 8 5

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A “chemical change” is a change that produces matter with a different composition than the original matter. Chemical Change A change in which

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one or more substances are converted into different

substances. Heat and light are often evidence of a chemical change.

Properties of Compounds Quite different properties than their component elements.

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