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Mass or weight 3Gram (g) = weight of 1 cubic centimeter (cm) of water at 4°C and 760 mm Hg Mass = quantity of material Weight = mass x gravitational force Length Meter (m) =100 cm =1000 millimeters (mm) =39.37 in. Volume Liter (L) =volume of 1 kilogram (kg) of H. 2.

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Dissolve the samples in 100mL of hot deionized water and add 2 drops of phenolphthalein. Use the pH meter to record the initial pH of the solution. Add more than sufficient amounts of NaOH to the vitamin C samples (~10mL excess) and titrate the excess NaOH with the 0.1M HCl standard solution using a pH meter.

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Determine the mass of the cylinder using the centigram balance.
Calculate the density of the metallic cylinder. (Recall that $d=m/V$.)
Calculate the Percent Error in your density determination knowing that the theoretical values of the cylinder densities are:
Brass:.....8.50 g/cm³.
Copper:.....8.96 g/cm³.

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