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Chemistry Inquiry Answers Chem Quest 53 concept of significant figures has the answer. When multiplying and dividing numbers, you need to round your answers to the correct number of significant figures. To round correctly, follow these simple steps: 1) Count the number of significant figures in each number. 2) Round your

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answer to the least number of

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X = chemical symbol of the element Z = [atomic number] A =

[mass number] , , and are notations that represent isotopes of carbon. H, and are notations that represent isotopes of hydrogen.

The part of the atom where the protons and neutrons are is called the nucleus. A X Z 12 C 6 C 13 6 C 14 6 12C 6 C 13 6 C 14 6 1H 1 H 2 1 1 H 1 2 H 1 3 3 1

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During a chemical reaction, new substances are formed. Reactants are transformed into different products. Atoms are never created or destroyed, but they are rearranged. A chemical equation represents

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what happens during a reaction. The following is an example of a chemical equation: Example Equation: $\text{Ca} + \text{HNO}_3 \rightarrow \text{Ca}(\text{NO}_3)_2 + \text{H}_2$

ChemQuest 28 - Weebly

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Important Questions And Answers helps students of class 11 and also for students studying for various competitive exams. Students are advised to practice and understand all the questions accordingly.

1.

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