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a diprotic acid. A
student determined
the value of x in the
formula of hydrated
ethanedioic acid, $\text{HOOC} - \text{COOH} \cdot x\text{H}_2\text{O}$,

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by titrating a known mass of the acid with a $0.100 \text{ mol dm}^{-3}$ solution of NaOH (aq). 0.795 g of ethanedioic acid was dissolved in distilled water and made up to a total ...

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: aluminium ions 3 : 1
amount Al 13 1 1.296
10 C 96500Cmol 3 «»
4.48 107 «mol» mass
Al « 4.48 107 mol
26.98 g mol 1 » 1.21
109 «g» Award [3] for
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studying are: #1:

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