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Answer Key CHAPTER 6 THERMOCHEMISTRY. 6.15
Recall that the work in gas expansion is equal to the product of the external, opposing pressure and the change in volume.

Chapter 6 Chemistry Answer Key

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The probability for all the particles to be on one side is $\frac{1}{32}$. This probability is noticeably lower than the $\frac{1}{8}$ result for the four-particle system. The conclusion we can make is that the probability for all the particles to stay in only one part of the system will decrease rapidly as the number of particles increases, and, for instance, the probability for all molecules of gas to ...

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energy changes (that is, energy produced or absor... A system that can exchange neither energy nor matter with its... Chemistry matter and change chapter 11 study guide answer key

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(a) (b) Answer is one of the following. A/yellow: mass = 65.14 kg, volume = 3.38 L, density = 19.3 kg/L, likely identity = gold. B/blue: mass = 0.64 kg, volume = 1.00 L, density = 0.64 kg/L, likely identity = apple.

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