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following reaction $K_c =$

0.513 at 500 K. $N_2O_4(g)$

$\rightleftharpoons 2 NO_2(g)$ If a reaction

vessel initially contains an

N_2O_4 concentration of

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0.0500 M at 500 K, what are the equilibrium concentrations of N_2O_4 and NO_2 at 500 K? a.

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