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BASIC PRINCIPLES

AND TECHNIQUES

(CH 20)

~~Nomenclature:~~

~~Functional~~

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13 - Properties  
of Solutions:  
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PLAY. A solution

is a.

homogeneous

mixture. The

first step of

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\_\_\_\_\_ thermal energy, whereas the mixing process \_\_\_\_\_

it. to absorb; release. A

student desires to remove the color of naphthalene (moth balls), a nonpolar aromatic

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hydrocarbon,

from his  
laboratory coat.

What solvent  
should he ...

Answers

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calorimetry.

heat capacity.

specific heat.

thermochemistry.

the study of

heat flow and

measurement. the

amount of heat

needed to raise

the temperature

of an objec...

heat capacity of

1 gram of a



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substance. the  
study of the  
changes in heat  
in chemical  
reactions.

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reaction rate

definition

average rate for

a time interval

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calculation of

order and  $k$  from

concentration

and rate factors

affecting rate

(and why)

concentrations

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of reactants,

temperature,

surface area,

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reaction

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elementary steps

rate laws for

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reactions

multistep

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determining step

activation

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to temperature  
relation to rate  
energy diagrams  
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complex ...

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can burn.

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science deal  
with  
composition,  
structure, and  
properties of

ma... is the  
study of

substances that  
contain the  
element

carbon. All... the  
mixture of which  
bond make  
organic



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carbon and  
hydrogen are  
organic  
substances.

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12 - States of

Matter Section

12.1 Gases 1.

motion 2. a.

small b. forces

c. random d.

elastic; kinetic

3. KE  $\frac{1}{2}mv^2$  4.

Temperature 5.

true 6. true 7.

false 8. true 9.

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Chapter 10. false

11. true 12.

false 13. a 14.

a 15. d 16. d

17. b 18. b 19.

b 20. barometer

21 . . .

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12 SOLUTIONS

MANUAL 30.

Compare and  
contrast

sublimation and  
evaporation. In  
both processes,  
the substances  
become a vapor.

During

sublimation, the

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substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

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deep into the

study of matter

and its related

properties. The

first chapter is

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where students

will learn about

the solid state

of matter in

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students will  
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mixture.

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begins with the

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nomenclature of

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phenols and

ethers. They

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of alcohols,  
phenols, and  
ethers. They  
further explore  
the properties  
of the same.

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Ketones and

Carboxylic Acids

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decide whether  
you will use it  
or some  
variation on it  
to complete the

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tronegativity,

Types of

Chemical Bonds,

and Bond

Polarity

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following. Be

sure to check

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determine how

much you are

expected to know

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