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questions and calculation. Read the
footnote when you see an item marked
with a clearly show all mathematical
work in equation editor where needed.

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Unknown code Volume of base to reach equivalence point pH at equivalence point Volume of base halfway to equivalence point Chem 1212 Lab Manual Revised 062015 pH at halfway to equivalence point Experimental pK_a of Unknown Experimental K_a of Unknown Which is the stronger acid (Acetic Acid or the

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cations: Ag^+ , K^+ and Fe^{3+} . You first add
a few drops of 1 M HCl to the
mixture. a. What will you observe?

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