

### Chapter 6 Covalent Bonding Answers

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CHAPTER 6 REVIEW Chemical Bonding SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. a A chemical bond between atoms results from the attraction between the valence electrons and of different atoms. (a) nuclei (c) isotopes (b) inner electrons (d) Lewis structures 2. b A covalent bond consists of (a) a shared electron.

#### [6 Chemical Bonding](#)

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Section 6.2 – Covalent Bonding. A covalent bond is a chemical bond in which two atoms share a pair of valence electrons. When two atoms share one pair of electrons, the bond is called a single bond.

#### [Chapter 6 – Chemical Bonds](#)

Covalent Bond is a Is a chemical bond in which two atoms share valence electrons Which ones show orbitals of atoms overlapping when a covalent bond forms? [ electrons dot, structural formula, space -filling, electron cloud]?

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