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800 Chapter 25 Types of Radiation

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Discuss Explain that the nuclei of a radioactive element spontaneously decompose. Nuclear chemistry is the study of changes in matter that originate in atomic nuclei. Ask, What types of radiation exist, and how harmful are they? (The three most common types of radiation emitted by unstable nuclei are

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#### ~~25.1 Nuclear Radiation 25~~

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SECTION 25.3 FISSION AND FUSION  
OF ATOMIC NUCLEI (pages 810–813)

This section describes nuclear fission and nuclear fusion. It discusses their potential as sources of energy, methods used to

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control them, and issues involved in  
containment of nuclear waste. Nuclear  
Fission (pages 810–811) 1.

~~SECTION 25.1 NUCLEAR RADIATION~~  
~~(pages 799–802)~~

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this chapter you will study nuclear chem-

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istry, which is concerned with the structure of atomic nuclei and the changes they undergo. An application of a nuclear reaction is shown in the photo of the human neck and skull. Table 25-1 offers a comparison of chemical and nuclear reactions. Chapter 25: Nuclear Chemistry

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unstable nucleus releases energy. 25.1.2  
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Terms in this set (37) Neutron Ejection.

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when a neutron is emitted from the nucleus.  ${}^1_0\text{n}$ . Particle for Neutron Ejection.  
 ${}^4_2\text{He} \rightarrow {}^1_0\text{n} + {}^3_2\text{He}$ . Chapter 25 - Nuclear Chemistry Flashcards | Quizlet

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Chapter 25: Nuclear Chemistry - Jayne

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Answers Nuclear Reactions □ Nuclear reactions involve changes in the nucleus, whereas chemical reactions involve the loss, gain, and sharing of electrons. □

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Different isotopes of the same element may undergo very different nuclear reactions,

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Objectives 25.2.1 ...



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~~te\_chapter\_25\_2\_1.pdf 25.2 Nuclear Transformations ...~~

Nuclear Reactions □ Nuclear reactions involve changes in the nucleus, whereas chemical reactions involve the loss, gain, and sharing of electrons. □ Different isotopes of the same element may undergo very different nuclear reactions, even

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though an element's isotopes all share the same chemical characteristics.

### ~~PowerPoint Chapter 18: Nuclear Chemistry~~

Chemistry End of Chapter Exercises.

Write a brief description or definition of each of the following: (a) nucleon (b)  $\alpha$

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particle (c)  $\alpha$  particle (d) positron (e)  $\gamma$  ray  
(f) nuclide (g) mass number (h) atomic  
number. Which of the various particles ( $\alpha$   
particles,  $\beta$  particles, and so on) that may  
be produced in a nuclear reaction are  
actually ...

~~21.2 Nuclear Equations~~  $\square$  ~~Chemistry~~

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