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Effusion rates for nitrogen (N_2) and neon (Ne). Rate N_2 /Rate Ne = 0.849

Rate N_2 = Rate Ne \times molar mass Ne / molar mass N_2

$0.849 = \frac{\text{Rate Ne}}{\text{Rate } N_2} \times \frac{20.18}{28.02}$

Calculate the ratio of diffusion rates for carbon monoxide and carbon dioxide.

1.25 Rate CO = Rate CO₂ \times

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