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## Chapter 11 Review Gases Section 4 Answers

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SECTION 1 Date CHAPTER 11 REVIEW Gases Class SHORT ANSWER Answer the following questions in the space provided. b Pressure - orce For a constant force, when the surface area is tripled the surface area pressure is (a) doubled. as much. (c ripled. 7-0 (d) unchanged. Rank

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the following pressures in increasing order. (c) 76 torr (a) 50 kPa O,  
OOÏctbv-x

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If a gas and a liquid are the same temperature and pressure, diffusion occurs much faster in the gas because. A. there are more elastic collisions between the particles in a gas. B. gases are more compressible. C. the particles move faster in a gas and there is a greater distance between them.

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Section 11.4 Dalton's Law of Partial Pressures Goals To describe the properties of mixtures of gases. To describe calculations that deal with mixtures of gases. In the real world, gases are usually mixtures. This section describes how mixing gases affects the properties of the resulting mixture.

*Chapter 11 - Gases*

462 Chapter 11 Gases Discovering the Relationships Between Properties  
If we want to explain why a weather balloon carrying instruments into the upper atmosphere expands as it rises, we need to consider changes in the properties of the gases (pressure, volume, temperature, or

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number of gas particles) inside and outside the balloon.

*Chapter 11 Gases - An Introduction to Chemistry*

CHAPTER 11 REVIEW Gases SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. b Pressure surface area. For a constant force, when the surface area is tripled the pressure is (a) doubled. (b) a third as much. (c) tripled. (d) unchanged. 2. d, c, a, b Rank the following pressures in increasing order. (a) 50 kPa (c) 76 torr (b) 2 atm (d) 100 N/m<sup>2</sup> 3.

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this theory explains some of the properties of ideal gases. In this chapter, you will study the predictions of kinetic-molecular theory for gases in more detail. This includes the relationship among the temperature, pressure, volume, and amount of gas in a sample. SECTION 11.1 Key Terms pressure newton barometer millimeters of mercury

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*SECTION 11.1 Gases and Pressure - Pickford High School*

CHAPTER 11 REVIEW Gases SECTION 2 SHORT ANSWER Answer the following questions in the space provided. 1. State whether the pressure of a fixed mass of gas will increase, decrease, or stay the same in the following circumstances: increase a. temperature increases, volume stays the same decrease b. volume increases, temperature stays the same

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discharges were observed, as well as the effects of small amounts of tungsten in the argon discharge. These topics are followed by discussions on the verification of the Langmuir Probe-Theory; the disturbing effects of mercury vapor blast from the cathode spot on the mercury pool; and the phenomena observed in the low pressure discharges. Other chapters explore the Faraday dark space of the glow discharge and how the electrons of a beam in a plasma are given their higher velocity components. The final chapters introduce the concept of plasma oscillation. This book will be of value to electronics engineers and technical workers.

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