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Match description on the right to the

correct crystal type on the left Modern

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of matter section 1. b ionic crystal (a) has

mobile electrons in the crystal c covalent

molecular crystal (b) is hard, brittle, and

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nonconducting a metallic crystal (c) typically has the lowest melting point of the four

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CHAPTER 10 REVIEW States of Matter

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the following questions in the space

provided. 1. a When a substance in a

closed system undergoes a phase change

and the system reaches equilibrium, (a) the

two opposing changes occur at equal rates.

(b) there are no more phase changes. (c)

one phase change predominates.

10 States of Matter - Ms. Agostine's

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Theory The kinetic-molecular theory of
matter is based on the idea that particles of
matter are always in motion and this
motion has consequences that affect its
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