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chapter 2 | GCSE Physics -

Factors Affecting Gas Pressure

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part 4 | Class 11 Physics |
Kinetic Theory of Gases | #2

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Behaviour of Real Gases | For JEE
& NEET Behavior Of Gases
Review 2

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- Gases are mostly empty space
- Gases diffuse and mix rapidly

Toowong

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- The properties of a gas are

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Behavior Of Gases Review 2 Answers - Maharashtra

2. Gas particles are in random, constant, straight line motion. 3. When particles collide with each other (or any container), the collisions are said to be elastic. 4.

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The volume that gas particles take up is negligible. And from the distance between particles is relatively great.

Behavior of Gases You'll

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BEHAVIOR OF GASES REVIEW
Page 102 Chemistry Unit

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Assessment 2007 Baltimore County Public Schools 11.

Calculate the new temperature of a gas when 1500 mL at 25°C is suddenly compressed to 500 mL.

Charles' Law $\frac{K \text{ mL}}{K} = \frac{K \text{ mL}}{K}$ $\frac{V_1 T_1}{V_2 T_2}$
 $\frac{1500 (298)}{500 (T_2)} = \frac{1 (T_1)}{1 (T_2)}$

12. A flask contains 34.6 kPa of

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CO 2

Student Review Packet Answer
Key

Behavior Of Gases Review 2

SECTION 2 BEHAVIOR OF

GASES 1. a measure of how fast
the particles of an object are

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moving 2. when it is heated 3.
Temperature of gas particles
Energy of gas particles Volume of
gas particles 1) 20 ° C Particles
have the smallest amount of
energy. Volume is smallest. 2)
50 ° C Particles have more

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Answers properties of gases. The imprecision is known as the non-ideal behavior of gas, and the van der Waals equation. $(P + \frac{n^2a}{V^2})(V - nb) = nRT$. has been

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introduced to deal with non-ideal
behavior of gases in Ideal gas law.

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SECTION 2 BEHAVIOR OF

GASES 1. a measure of how fast
the particles of an object are
moving 2. when it is heated 3.

Temperature of gas particles

Energy of gas particles Volume of
gas particles 1) 20 ° C Particles

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have the smallest amount of
energy. Volume is smallest. 2)
50 ° C Particles have more

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SECTION 2 BEHAVIOR OF

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Temperature of gas particles
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energy. Volume is smallest. 2)
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Answers

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SECTION 2 BEHAVIOR OF

GASES 1. a measure of how fast the particles of an object are moving 2. when it is heated 3.

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Temperature of gas particles
Energy of gas particles Volume of
gas particles 1) 20 ° C Particles
have the smallest amount of
energy. Volume is smallest. 2)
50 ° C Particles have more energy

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You usually cannot feel it, but air has pressure. The gases in Earth ' s atmosphere exert pressure against everything they contact. The atmosphere rises

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high above Earth ' s surface. It contains a huge number of individual gas particles. As a result, the pressure of the tower of air above a given spot on Earth ' s surface is substantial.

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