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Tips to Balance Chemical Equations Tip # 1: When you are trying to balance the chemical equations, you should remember that you can only change the value... Tip # 2: You

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should remember that polyatomic ions should be balanced as a whole. For instance, SO₄ should be balanced... Tip # 3: You should ...

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Balancing Chemical Equations – Answer Key Balance the

equations below: 1) $1 \text{ N}_2 + 3 \text{ H}_2 \rightarrow 2 \text{ NH}_3$ 2) $2 \text{ KClO}_3 \rightarrow 2$

$\text{KCl} + 3 \text{ O}_2$ 3) $2 \text{ NaCl} + 1 \text{ F}_2 \rightarrow 2 \text{ NaF} + 1 \text{ Cl}_2$ 4) $2 \text{ H}_2 + 1 \text{ O}_2$

$\rightarrow 2 \text{ H}_2\text{O}$ 5) $1 \text{ Pb(OH)}_2 + 2 \text{ HCl} \rightarrow 2 \text{ H}_2\text{O} + 1 \text{ PbCl}_2$ 6) 2

$\text{AlBr}_3 + 3 \text{ K}_2\text{SO}_4 \rightarrow 6 \text{ KBr} + 1 \text{ Al}_2(\text{SO}_4)_3$ 7) $1 \text{ CH}_4 + 2 \text{ O}_2 \rightarrow$

$1 \text{ CO}_2 + 2 \text{ H}_2\text{O}$ 8) $1 \text{ C}_3\text{H}_8 + 5 \text{ O}_2 \rightarrow 3 \text{ CO}_2 + 4 \text{ H}_2\text{O}$ 9) 2

$\text{C}_8\text{H}_{18} + 25 \text{ O}_2 \rightarrow 16 \text{ CO}_2 + 18 \text{ H}_2\text{O}$ 10) $1 \text{ FeCl}_3 + 3 \text{ NaOH}$

$\rightarrow 1 \text{ Fe(OH)}_3 + 3 \text{ NaCl}$ 11) $4 \text{ P} + 5 \text{ O}_2 \rightarrow 2 \text{ P}_2\text{O}_5$

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To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in “H₂+O₂” in the Reactants box and “H₂O” in the Products box. This...

~~Student Exploration—Balancing Chemical Equations
(ANSWER KEY)~~

Balancing chemical equations? | Yahoo Answers. Balancing chemical equations? write and balance each of the following equations, and then identify each by type. a) hydrogen+iodine---> hydrogen iodide. b) lithium+hydrochloric acid---> lithium chloride +hydrogen. <https://answers.yahoo.com/question/index?qid=20120202133555AAAtGiqB>

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This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating

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molecules below the initial reaction: $H_2 + O_2 \rightarrow H_2O$. 1. How many atoms are in a hydrogen molecule (H_2)? 2 2. How many atoms are in an oxygen molecule (O_2)? 2 3. How many hydrogen and oxygen atoms are in a water molecule (H_2O)? H: 2 O: 1 4. In general, what does a subscript (such as the "2" in H

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Anne Marie Helmenstine, Ph.D. Updated January 04, 2019. A balanced chemical equation gives the number and type of atoms participating in a reaction, the reactants, products, and direction of the reaction. Balancing an unbalanced equation is mostly a matter of making certain mass and charge are

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balanced on the reactants and products side of the reaction arrow.

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Balancing Equations: Answers to Practice Problems. 1.

Balanced equations. (Coefficients equal to one (1) do not need to be shown in your answers). (a) $2\text{Fe} + 3\text{Cl}_2 \rightarrow 2\text{FeCl}_3$.

(b) $4\text{Fe} + 3\text{O}_2 \rightarrow 2\text{Fe}_2\text{O}_3$. (c) $2\text{FeBr}_3 + 3\text{H}_2\text{SO}_4 \rightarrow$

$1\text{Fe}_2(\text{SO}_4)_3 + 6\text{HBr}$ (d) $1\text{C}_4\text{H}_6\text{O}_3 + 1\text{H}_2\text{O} \rightarrow 2\text{C}_2\text{H}_4\text{O}_2$.

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To balance a chemical equation, enter an equation of a

chemical reaction and press the Balance button. The

balanced equation will appear above. Use uppercase for the

first character in the element and lowercase for the second

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character. Examples: Fe, Au, Co, Br, C, O, N, F.

~~Chemical Equation Balancer~~

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Start level 2 of the balancing equation game. Take turns in your group to balance the equations in the sim, using your strategies from Level 1, and adding new strategies as

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needed. Each person should be in charge of balancing at least one equation, asking for help from the group as needed. As a group, write down the equations as you solve them.

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